

## Aieee 2010 Solution

Aieee 2010 Solution AIEEE-2010-CODE-A 5 (3) ultraviolet rays (4) microwaves. Key. (2) Sol.: P 4000 current leads the voltage by  $30^\circ$ . The power  $20 E$   
10 hc hc 10 hc 17 4000 4 3 10 6.6 108 34 17 4 9 19.8 10 4 4.9 10 9 49 10 10 49A 0.1A 100A o it is X-ray (2) is correct. 16. AIEEE - 2010 S O L U T I O  
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cut idea of the difficulty level, the pattern and the distribution of questions across various topics. JEE Main/Advanced 2010 Paper Solutions -  
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aldehyde having a molecular mass of 44 u. The alkene is (1) propene (2) 1-butene (3) 2-butene (4) ethene From amongst the following alcohols the  
one that would react fastest with conc. HCl and ... AIEEE 2010 with Solutions - Notes For Free AIEEE 2010 Physics (part 1) by Resonance #AIEEE2010  
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On mixing, heptane and octane form an ideal solution. At 373 K, the vapour pressures of the two liquid components (heptane and octane) are 105  
kPa and 45 kPa respectively. AIEEE 2010 | Solutions Question 13 | Chemistry | JEE Main ... AIEEE-2010 CHEMISTRY, PHYSICS & MATHS PART-A :  
CHEMISTRY 1. The standard enthalpy of formation of  $NH_3$  is  $-46.0 \text{ kJ mol}^{-1}$ . If the enthalpy of formation of  $H_2$  from its atoms is  $-436 \text{ kJ mol}^{-1}$  and  
that of  $N_2$  is  $-712 \text{ kJ mol}^{-1}$ , the average bond enthalpy of N-H bond in  $NH_3$  is 1)  $-964 \text{ kJ mol}^{-1}$  2)  $+352 \text{ kJ mol}^{-1}$  3)  $+1056 \text{ kJ mol}^{-1}$  4)  $-1102 \text{ kJ}$   
 $\text{mol}^{-1}$  2. AIEEE-2010 CHEMISTRY, PHYSICS & MATHS AIEEE-2010 QUESTION ... IITJEE2010-Paper 2-CMP-1 FIITJEE Ltd., FIITJEE House, 29-A, Kalu Sarai,  
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PHYSICS & MATHS 51. Two identical charged spheres are suspended by strings of equal lengths. The strings make an angle of  $30^\circ$  with each other.  
When suspended in a liquid of density  $0.8 \text{ g cm}^{-3}$ , the angle remains the same. AIEEE-2010 QUESTION PAPER - MAFIADOC.COM 18. A binary liquid  
solution is prepared by mixing n-heptane and ethanol. Which one of the following statements is correct regarding the behaviour of the solution? (1)  
The solution is non-ideal, showing +ve deviation from Raoult's Law (2) The solution is non-ideal, showing -ve deviation from Raoult's Law Solutions  
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2010 25 04 10 Questions & Solutions 5 www. prernaclases.com. 52. For a particular reversible reaction at temperature T,  $\Delta H$  and  $\Delta S$  were found to be both +ve. If  $T_e$  is the temperature at equilibrium, the reaction would be spontaneous when AIEEE 2010 Chem solution by prerna classes - [PDF Document] aieee 2010 In aqueous solution the ionization constants for carbonic acid are  $K_1 = 4.2 \times 10^{-7}$  and  $K_2 = 4.8 \times 10^{-11}$  Select the correct statement for a saturated 0.034 M solution of the carbonic acid. AIEEE 2010 | Equilibrium Question 13 | Chemistry | JEE ... JEE Main (AIEEE) 2010 Question Paper. ... 33. A solution containing 2.675 g of  $\text{CoCl}_3 \cdot 6\text{NH}_3$  (molar mass = 267.5 g mol<sup>-1</sup>) is passed through a cation exchanger. The chloride ions obtained in solution were treated with excess of  $\text{AgNO}_3$  to give 4.78 g of  $\text{AgCl}$  (molar mass = 143.5 g mol<sup>-1</sup>). The formula of the complex is (At).

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